

Revision date: 2009/10/02 Page: 1/6
Version: 1.0 (30338188/SDS\_GEN\_US/EN)

# 1. Product and Company Identification

Company
BASF CORPORATION
100 Campus Drive
Florham Park, NJ 07932, USA

24 Hour Emergency Response Information CHEMTREC: 1-800-424-9300 BASF HOTLINE: 1-800-832-HELP

### 2. Hazards Identification

## **Emergency overview**

CAUTION:

HARMFUL IF SWALLOWED.

Ingestion may cause irritation to mucous membranes.

May cause skin irritation.

Avoid contact with the skin, eyes and clothing.

Wash thoroughly after handling. Keep container tightly closed.

State of matter: liquid Colour: yellow Odour: sweetish

## Potential health effects

## Primary routes of exposure:

Routes of entry for solids and liquids include eye and skin contact, ingestion and inhalation. Routes of entry for gases include inhalation and eye contact. Skin contact may be a route of entry for liquified gases.

# 3. Composition / Information on Ingredients

CAS NumberContent (W/W)Chemical name10124-37-5>= 10.0 - <= 30.0 %</td>calcium nitrate

### 4. First-Aid Measures

#### General advice:

First aid personnel should pay attention to their own safety. Remove contaminated clothing.

#### If inhaled

If difficulties occur after vapour/aerosol has been inhaled, remove to fresh air and seek medical attention.

Revision date: 2009/10/02 Page: 2/6
Version: 1.0 (30338188/SDS GEN US/EN)

#### If on skin:

After contact with skin, wash immediately with plenty of water and soap. Under no circumstances should organic solvent be used. If irritation develops, seek medical attention.

#### If in eyes:

Wash affected eyes for at least 15 minutes under running water with eyelids held open, consult an eye specialist.

#### If swallowed:

Rinse mouth immediately with water. Seek medical attention if necessary. Do not induce vomiting unless told to by a poison control center or doctor.

# 5. Fire-Fighting Measures

Flash point: Non-flammable. Autoignition: Non-glammable not applicable

#### Suitable extinguishing media:

foam, water spray, dry extinguishing media, carbon dioxide

### Unsuitable extinguishing media for safety reasons:

water jet

#### Hazards during fire-fighting:

carbon monoxide, carbon dioxide, harmful vapours, nitrogen oxides, fumes/smoke, carbon black

#### Protective equipment for fire-fighting:

Wear a self-contained breathing apparatus.

#### Further information:

The degree of risk is governed by the burning substance and the fire conditions. Contaminated extinguishing water must be disposed of in accordance with official regulations.

## 6. Accidental release measures

#### Personal precautions:

Use personal protective clothing. Do not breathe vapour/aerosol/spray mists. Handle in accordance with good building materials hygiene and safety practice.

#### **Environmental precautions:**

Contain contaminated water/firefighting water. Do not discharge into drains/surface waters/groundwater.

#### Cleanup:

For small amounts: Pick up with inert absorbent material (e.g. sand, earth etc.). Dispose of contaminated material as prescribed.

For large amounts: Pump off product.

## 7. Handling and Storage

#### **Handling**

#### General advice:

Avoid aerosol formation. Avoid inhalation of mists/vapours. Avoid skin contact. No special measures necessary provided product is used correctly.

## Protection against fire and explosion:

The product does not contribute to the spreading of flames, nor is it self combustible, not explosive. Take precautionary measures against static discharges.

Revision date: 2009/10/02 Page: 3/6 Version: 1.0 (30338188/SDS GEN US/EN)

Storage

#### General advice:

Keep only in the original container in a cool, well-ventilated place. Protect from direct sunlight. Store protected against freezing.

### Temperature tolerance

Protect from temperatures below: 0 °C

The packed product must be protected from temperatures below the indicated one.

# 8. Exposure Controls and Personal Protection

### Personal protective equipment

#### Respiratory protection:

When workers are facing concentrations above the occupational exposure limits they must use appropriate certified respirators.

#### Hand protection:

Wear chemical resistant protective gloves.

#### Eye protection:

Safety glasses with side-shields.

### **Body protection:**

Impermeable protective clothing

#### General safety and hygiene measures:

Avoid contact with the skin, eyes and clothing. In order to prevent contamination while handling, closed working clothes and working gloves should be used. Handle in accordance with good building materials hygiene and safety practice. When using, do not eat, drink or smoke. Hands and/or face should be washed before breaks and at the end of the shift. At the end of the shift the skin should be cleaned and skin-care agents applied. Gloves must be inspected regularly and prior to each use. Replace if necessary (e.g. pinhole leaks).

# 9. Physical and Chemical Properties

Form: liquid

Odour: sweetish, slight odour

Colour: yellow

pH value: approx. 9.0 (20 °C)

Freezing point: approx. -4 °C
Boiling point: approx. 108 °C

approx. 226 °F

Information on: water

Vapour pressure: 23 mbar (20 °C)

Density: 1.29 g/cm3 (20 °C) Relative density: approx. 1.29 (20 °C)

Partitioning coefficient n
The value has not be determined because

octanol/water (log Pow): the substance is inorganic.

Solubility in water: soluble

# 10. Stability and Reactivity

# Conditions to avoid:

Avoid extreme temperatures.

#### Substances to avoid:

nitrogen oxides

Revision date: 2009/10/02 Page: 4/6

Version: 1.0 (30338188/SDS\_GEN\_US/EN)

#### **Hazardous reactions:**

The product is stable if stored and handled as prescribed/indicated.

#### **Decomposition products:**

nitrogen oxides

#### Thermal decomposition:

No decomposition if stored and handled as prescribed/indicated.

# Oxidizing properties:

not fire-propagating

# 11. Toxicological information

### **Acute toxicity**

Information on: Calcium nitrite Assessment of acute toxicity:

Of moderate toxicity after single ingestion. There is a risk of damage to the blood (methemoglobinemia) after a

single uptake of large quantities.

#### Oral:

Information on: calcium nitrate

Type of value: LD50

Species: rat

Value: 3,430 mg/kg (BASF-Test)

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# Repeated dose toxicity

Information on: calcium nitrate

Assessment of repeated dose toxicity:

The substance may cause damage to the hematological system after repeated ingestion.

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# 12. Ecological Information

## **Aquatic toxicity**

Information on: Calcium nitrite Assessment of aquatic toxicity:

The product has not been tested. The statement has been derived from products of a similar structure and

composition.

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#### Fish

Information on: Calcium nitrite

# 13. Disposal considerations

#### Waste disposal of substance:

Recommendations: Use excess product in an alternate beneficial application. Do not discharge into drains/surface waters/groundwater. Dispose of in accordance with local authority regulations.

Revision date : 2009/10/02 Page: 5/6 Version: 1.0 (30338188/SDS\_GEN\_US/EN)

#### Container disposal:

Contaminated packaging should be emptied as far as possible; then it can be passed on for recycling after being thoroughly cleaned.

# 14. Transport Information

Land transport

USDOT

Not classified as a dangerous good under transport regulations

Sea transport

**IMDG** 

Not classified as a dangerous good under transport regulations

Air transport IATA/ICAO

Not classified as a dangerous good under transport regulations

# 15. Regulatory Information

**Federal Regulations** 

Registration status:

Chemical TSCA, US released / listed

**OSHA hazard category:** No data available.;

EPCRA 311/312 (Hazard categories): Acute; Chronic

**EPCRA 313:** 

<u>CAS Number</u> <u>Chemical name</u> 10124-37-5 calcium nitrate

State regulations

State RTKCAS NumberChemical nameNJ10124-37-5calcium nitrate

## 16. Other Information

NFPA Hazard codes:

Health: 2 Fire: 0 Reactivity: 0 Special:

**HMIS III rating** 

Health: 2<sup>m</sup> Flammability: 0 Physical hazard: 0

NFPA and HMIS use a numbering scale ranging from 0 to 4 to indicate the degree of hazard. A value of zero means that the substance possesses essentially no hazard; a rating of four indicates extreme danger. Although similar, the two rating systems are intended for different purposes, and use different criteria. The NFPA system was developed to provide an onthe-spot alert to the hazards of a material, and their severity, to emergency responders. The HMIS system was designed to communicate workplace hazard information to employees who handle hazardous chemicals.

Revision date: 2009/10/02 Page: 6/6 Version: 1.0 (30338188/SDS GEN US/EN)

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### **Local Contact Information**

BASF Construction Chemicals bcc\_prps@basf.com

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